

## Rapid report

Transport energetics of the  $\text{Cl}^-$  pump in *Aplysia* gut

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Received 23 July 1997; accepted 31 July 1997

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**Abstract**

Basolateral membranes of *Aplysia* foregut epithelia contain an ATP-dependent  $\text{Cl}^-$  transporter ( $\text{Cl}^-$  pump). Increased activity of the  $\text{Cl}^-$  pump, coupled to apical and basolateral membrane depolarization, changed the  $\text{Cl}^-$  transport energetics across the apical membrane but did not change the vectorially-opposite  $\text{Cl}^-$  transport energetics across the basolateral membrane. © 1997 Elsevier Science B.V.

**Keywords:** Gut; Active transport;  $\text{Cl}^-$  energetics; (*Aplysia*)

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Active  $\text{Cl}^-$  absorption by the *Aplysia californica* (seahare) foregut is mediated by a basolaterally-located  $\text{Na}^+$ -independent, electrogenic,  $\text{Cl}^-$ -stimulated ATPase [1,2]. This  $\text{Cl}^-$  pump accounts for the intracellular  $\text{Cl}^-$  electrochemical potential ( $\bar{\mu}$ ) being less than the extracellular  $\text{Cl}^-$   $\bar{\mu}$  [3].  $\text{Cl}^-$  transport across the apical membrane into the cytosol of these cells is mediated via voltage-regulated channels [4]. In addition, aminoisobutyric acid (AIB), a nonmetabolizable amino acid, is actively accumulated by the *Aplysia* foregut in the presence of  $\text{Na}^+$  [5]. This  $\text{Na}^+$ -dependent AIB transport enhanced the net mucosal-to-serosal  $\text{Cl}^-$  flux across the *Aplysia* foregut [5]. In view of this apical membrane coupling of  $\text{Na}^+$ , AIB and  $\text{Cl}^-$ , the present study was undertaken to determine whether this event altered the energy gradient across the basolateral membrane where the  $\text{Cl}^-$  pump transports  $\text{Cl}^-$  uphill.

Adult seahares (*Aplysia californica*) were ob-

tained from Marinus (Westchester, CA) and were maintained at 25°C in circulating filtered sea water. The animals were sacrificed and their posterior foreguts were removed, slit longitudinally, rinsed and then positioned between two halves of a Lucite chamber described previously [3] which allowed measurement of transepithelial electrical potential ( $\Psi_{\text{ms}}$ ) and, simultaneously, the introduction of microelectrodes into the surface epithelial cells. The chamber exposed the tissue to an oxygenated seawater medium. The formula for the seawater medium was (in mM): NaCl, 462.0;  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , 2.4; KCl, 10.0;  $\text{KHCO}_3$ , 2.4;  $\text{MgCl}_2$ , 9.8;  $\text{CaCl}_2$ , 11.4. The total osmolality of the bathing medium was 1000 mosmol/l and the final pH was 7.8. Microelectrodes for measurements of mucosal membrane potential ( $\Psi_{\text{m}}$ ) and  $\bar{\mu}_{\text{Cl}}$  were constructed and utilized as previously described [3]. Briefly, the experimental protocol was as follows: after the excised tissue was placed into cells lining the gut villi to obtain an independent estimate of  $\Psi_{\text{m}}$ , after which,  $\text{Cl}^-$ -selective microelectrodes were passed into the villus ep-

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ithelial cells to measure the intracellular  $\bar{\mu}_{\text{Cl}^-}$ . The intracellular  $\text{Cl}^-$  activity ( $a_{\text{Cl}^-}^i$ ) was calculated using

$$a_{\text{Cl}^-}^i = a_{\text{Cl}^-}^o / e^{2.303[(\Psi_i - \Psi_m) - \Psi'']/S} \quad (1)$$

where  $\Psi_i$  is the potential of the  $\text{Cl}^-$  electrode in the cell,  $a_{\text{Cl}^-}^i$  the activity of  $\text{Cl}^-$ ,  $\Psi''$  the potential of the  $\text{Cl}^-$  electrode in 500 mM NaCl,  $\Psi_m$  the mean mucosal membrane potential, and  $S$  is the slope of the electrode response defined as described previously [3,6]. In Table 1,  $\bar{\mu}_j$  is expressed as reversible work, in joules, required to transfer one equivalent of  $\text{Cl}^-$  across the mucosal or basolateral (serosal) membrane of a foregut villus epithelial cell and is calculated from

$$\bar{\mu}_j = RT \ln a_j^i / a_j^o + z\Psi F \quad (2)$$

where  $R$ ,  $T$ ,  $z$  and  $F$  have their usual physicochemical meanings and  $\Psi$  can either be  $\Psi_m$  or  $\Psi_s$  (basolateral or serosal membrane potential). The data obtained were analyzed statistically by Student's  $t$ -test.

As demonstrated in the present study (Table 1), mucosally-applied 80 mM AIB significantly depolarized both  $\Psi_m$  and  $\Psi_s$  and significantly hyperpolarized  $\Psi_{\text{ms}}$ . Therefore the calculated change in  $\Psi_s$  exceeded the empirically determined change in  $\Psi_m$ . Also seen in Table 1 are the findings that mucosally-applied AIB significantly increased  $a_{\text{Cl}^-}^i$  above control and significantly lowered the difference in  $\bar{\mu}_{\text{Cl}^-}$  across the basolateral membrane.

Mucosally-applied AIB caused a depolarization of  $\Psi_m$ , a hyperpolarization of  $\Psi_{\text{ms}}$  and an increase in  $a_{\text{Cl}^-}^i$  in the foregut cells of *Aplysia californica* (Table 1). These observations can be explained as follows: If an actively transported amino acid such as AIB is

$\text{Na}^+$ -coupled at the apical membrane [5], this would depolarize the  $\Psi_m$  because of the electrogenic or rheogenic nature of the mechanism [2,7]. This depolarization would then lower the electrochemical or thermodynamic energy gradient ( $\bar{\mu}_{\text{Cl}^-}^m$ ) opposing intracellular  $\text{Cl}^-$  entry from the mucosal bathing solution as shown in Table 1. This would then allow  $\text{Cl}^-$  to move across the mucosal membrane into the intracellular space with greater ease via the voltage-regulated  $\text{Cl}^-$  channels which were further opened by the  $\Psi_m$  depolarization [4]. The finding that  $a_{\text{Cl}^-}^i$  is significantly higher in the presence of mucosal AIB than in its absence (Table 1) lends strong support for this notion. The basolaterally-located electrogenic  $\text{Cl}^-$  pump or  $\text{Cl}^-$ -ATPase [1] would then accommodate the increased thermodynamic activity of  $\text{Cl}^-$  ( $\text{Cl}^-$  transport pool) by increasing its rate of work much as a variable output device which was described by Michaelis–Menten kinetics in an isolated  $\text{Cl}^-$  pump protein system [8]. In addition, this was previously demonstrated both by an increase in short-circuit current (SCC) and unidirectional mucosal-to-serosal  $\text{Cl}^-$  flux after mucosal AIB addition to the voltage-clamped, isolated foregut [5]. Since it has been shown that the major portion of the SCC before and after AIB addition to the mucosal solution is a  $\text{Cl}^-$  current [5], there would be a decrease in the negativity of  $\Psi_s$  by both the increased electrogenic  $\text{Cl}^-$  pump activity and the linkage of  $\Psi_m$  to  $\Psi_s$  through a low resistance extracellular shunt [9] which would lead to a greater serosally-negative  $\Psi_{\text{ms}}$  [10] as shown in Table 1. Whether AIB is present or absent in the mucosal bathing solution the thermodynamic energy gradient against which the  $\text{Cl}^-$  pump is moving  $\text{Cl}^-$  ( $\bar{\mu}_{\text{Cl}^-}^s$ )

Table 1

Potential profiles, intracellular  $\text{Cl}^-$  activities and transmembrane  $\text{Cl}^-$  electrochemical potential differences in *Aplysia californica* foregut bathed in NaCl seawater medium in the presence and absence of mucosal aminoisobutyric acid

|                     | $\Psi_m$ (mV)           | $\Psi_s$ (mV)           | $\Psi_{\text{ms}}$ (mV) | $a_{\text{Cl}^-}^i$ (mM) | $\bar{\mu}_{\text{Cl}^-}^m$ (J/equiv.) | $\bar{\mu}_{\text{Cl}^-}^s$ (J/equiv.) | $n$ |
|---------------------|-------------------------|-------------------------|-------------------------|--------------------------|--|--|-----|
| Before AIB addition | $-68.2 \pm 1.8$<br>(35) | $+67.9 \pm 1.4$<br>(35) | $-0.3 \pm 0.1$<br>(35)  | $8.6 \pm 0.5$<br>(22)    | $+7066 \pm 66$<br>(22)                 | $-7095 \pm 70$<br>(22)                 | 12  |
| After AIB addition  | $-61.1 \pm 1.8$<br>(35) | $+57.8 \pm 1.5$<br>(35) | $-3.3 \pm 0.4$<br>(35)  | $13.8 \pm 0.7$<br>(22)   | $+6663 \pm 55$<br>(22)                 | $-6981 \pm 48$<br>(22)                 | 12  |
|                     | $p < 0.01$              | $p < 0.01$              | $p < 0.01$              | $p < 0.01$               | $p < 0.01$                             | N.S.                                   |     |

Values are means  $\pm$  S.E. Numbers in parentheses are number of observations;  $n$  is the number of animals. Polarity of  $\Psi_m$  and  $\Psi_{\text{ms}}$  are relative to the mucosal solution. Polarity of calculated  $\Psi_s$  is relative to cytoplasm.  $a_{\text{Cl}^-}^i$  was calculated by means of Eq. (1).  $\bar{\mu}_{\text{Cl}^-}^m$  and  $\bar{\mu}_{\text{Cl}^-}^s$  were calculated by means of Eq. (2). (+) for  $\bar{\mu}_{\text{Cl}^-}^m$  represents downhill energy gradient from mucosal solution to cytosol. (–) for  $\bar{\mu}_{\text{Cl}^-}^s$  represents uphill energy gradient from cytosol to serosal solution.

does not change (Table 1). This is because in the presence of a maximally stimulating 80 mM AIB concentration [5], the increase in  $a_{\text{Cl}}^i$  (which favors a decrease in the energy gradient) almost equals the thermodynamic equivalent for the decrease in  $\Psi_s$  (which favors an increase in the energy gradient). These two forces (chemical and electrical potentials), which define the overall energy gradient against which the basolaterally-localized  $\text{Cl}^-$  pump must move  $\text{Cl}^-$ , quantitatively cancel one another because they are vectorially oriented in opposite directions and their respective quantitative changes are equivalent in the presence of AIB.

My thanks to Dr. William McD. Armstrong for helpful discussions during the course of the study. I would like to acknowledge the excellent technical assistance of F. Robbins. This investigation was supported by the Eppley Foundation for Research, Inc.

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